

Chapter 11 Review Gases Section 1 Worksheet Answers

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Chapter 11 Review Gases Section 4 Answers

CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided 1 The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol (c) multiplied by 224 L (b) divided by the mass of 1 mol (d) divided by 224 L 2 For the expression $V = nRT/P$

Chapter 11 Review Gases Section 4 Answers

CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b Pressure surf f a o c r e ce area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m²

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Modern Chemistry 97 Gases CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1

Chapter 11- Gases: Section 1: Gases and Pressure

Modern Chemistry 97 Gases CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. ____ The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2. ____ For the expression , P

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11.1 Gases and Their Properties 11.2 Ideal Gas Calculations 11.3 Equation Stoichiometry and Ideal Gases 11.4 Dalton's Law of Partial Pressures Convert between temperatures in the Celsius and Kelvin scales. (Section 2.6) Describe the particle nature of gases. (Section 3.1) Convert the amount of one substance in

Chapter 11 Review Gases Section

CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Pressure surface area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2.

Bing: Chapter 11 Review Gases Section

Chapter 11 183 Section 11.3 Equation Stoichiometry and Ideal Gases Goal: To show how gas-related calculations can be applied to equation stoichiometry problems. This section shows how we can combine calculations such as those found in Chapter 10 with the gas calculations described in Section 11.2 to do equation stoichiometry problems that

Chapter 11 Review Gases Mixed Answer Key

Chapter 11 Review Gases Mixed Chapter 11 181 Chapter 11 - Gases Review Skills 111 Gases and Their Properties Chapter 11 Map Chapter Checklist Read the Review Skills section If there is any skill mentioned that you have not yet neon gas mixed with argon gas (Objs 21 & 22) a If the total pressure of the mixture of gases is 130 kPa and the partial

Chapter 11 Gases - An Introduction to Chemistry

CHAPTER 11 REVIEW Gases SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (c) multiplied by 22.4 L. (b) divided by the mass of 1 mol. (d) divided by 22.4 L. 2.

Chapter 11 Review Gases Section 3 Short Answer

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CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances:
Chapter 11 Section 1 Gases and Pressure Objectives

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SECTION 3 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. c c The molar mass of a gas at STP is the density of that gas (a) multiplied by the mass of 1 mol. (b) divided by the mass of 1 mol. nRT (c) multiplied by 22.4 L. (d) divided by 22.4 L. For the expression $V =$ (a) increasing P

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